

Synthesis and Enzymatic Kinetic Resolution of α,α -Disubstituted Cyclic Hydroxy Nitriles

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Abstract: Herein, we describe the diastereoselective synthesis of five- and six-membered α,α -disubstituted cyclic β -hydroxy nitriles and their resolution via enzymatic transesterification. By this method, all possible stereoisomers were obtained in enantiopure form and high yield.

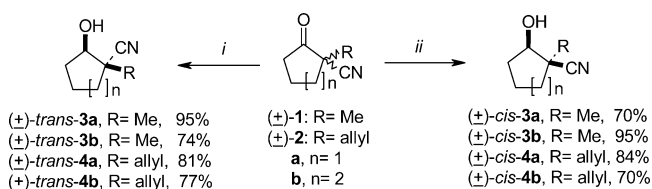
The asymmetric construction of quaternary carbon centers represents a challenging task in organic synthesis.¹ Despite numerous reports on bioreduction of ketones² and kinetic resolutions of secondary alcohols, very few strategies utilize these two methods to obtain fully substituted carbon atoms enantiomerically pure, and if so, usually with only moderate success.³

On the other hand, the importance of optically active β -hydroxy nitriles as suitable synthons for the preparation of γ -amino alcohols (like the antidepressant fluoxetine)⁴ is steadily growing. Thus, methodologies have been recently developed to prepare these alcohols via classical^{5a} or dynamic kinetic resolution,^{5b} reduction,^{5c} alkylation–reduction,^{5d} and addition processes.^{5e}

Therefore, we decided to combine both targets and chose racemic *cis*- and *trans*-1-alkyl-2-hydroxycycloalkane nitriles **3** and **4** for the kinetic resolution *via* enzymatic transesterification.

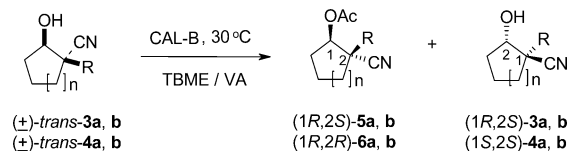
Keto nitriles **1** and **2** were synthesized *via* Thorpe–Ziegler reaction using the corresponding dinitriles followed by alkylation in α -position. After trying different reagents in the subsequent diastereoselective reduction (K-Selectride, Zn(BH₄)₂, CeCl₃/NaBH₄), Marcantoni's and Bartoli's procedure for the reduction of keto esters⁶ gave the best results. TiCl₄/BH₃·py was used as the chelating reagent to obtain the *trans* diastereomers and CeCl₃/

SCHEME 1^a



^a Reagents and conditions: (i) 1.5 equiv of TiCl₄, 1.5 equiv of BH₃·py, DCM, –100 °C, 1 h; (ii) 3.2 equiv of CeCl₃, 2 equiv of LiEt₃BH, THF, –100 °C, 2 h.

SCHEME 2



LiEt₃BH as the nonchelating reagent to obtain the *cis* diastereomers (Scheme 1).

Next, we carried out the resolution of racemic β -hydroxy nitriles *trans*-**3**, **4** by lipase *Candida antarctica* B (CAL-B)-catalyzed enantioselective acylation using a 3-fold excess of vinyl acetate (VA) as the acyl donor in *tert*-butyl methyl ether (TBME) at 30 °C (Scheme 2). These conditions were chosen since they gave the best results in our previous work on the resolution of cyclic β -hydroxy esters.⁷

Under these reaction conditions, *O*-acylation of the *trans* isomers (±)-**3a**, **3b**, **4a**, and **4b** took place smoothly, and 50% conversion could be reached, yielding both the substrates and the products in excellent enantiomeric excess (ee \geq 99%), corresponding to excellent enantioselectivities ($E > 200$) in all cases (Table 1). With respect to the ring size and the α -substituent, no difference in the enantioselectivities could be observed. However, CAL-B displayed a lower activity with R = allyl than with R = Me (compare entries 1, 2 with 3, 4). Furthermore, CAL-B reacted more sluggishly with the six-membered ring substrates in comparison to their five-membered ring analogues (compare entry 1 with 2, Table 1).

The same methodology was then applied to the *cis*-configured β -hydroxy nitriles (±)-**3** and (±)-**4** (Scheme 3). In all cases, substrates and products were isolated in very high yield and in excellent enantiomeric excess (ee \geq 99%, $E > 200$) after 2–13 h (Table 2). Again, no dependence of the enantioselectivity on the ring size was observed. However, a considerably lower activity of CAL-B toward (±)-*trans*-**4a**, **4b** in comparison to substrates (±)-*cis*-**4a**, **4b** was obtained (compare entries 3 and 4 in Tables 1 and 2).

The assignment of the relative configuration of alcohols **4a** and **4b** has been done on the basis of their ¹³C NMR and ¹H-NOESY spectra, as had been previously reported for substrates **3a** and **3b** (Figure 1).⁹ In the case of the

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TABLE 1. CAL-B-Catalyzed Enantioselective Acylation of (\pm)-*trans*- β -Hydroxy Nitriles

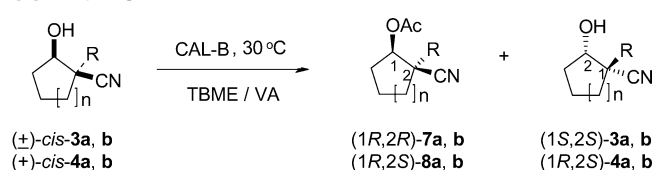
entry	<i>n</i>	R	substrate	<i>t</i> (h)	product		substrate		<i>c</i> ^c (%)	<i>E</i> ^c
					yield ^a (%)	ee ^b (%)	yield ^a (%)	ee ^b (%)		
1	1	Me	(\pm)- <i>trans</i> - 3a	4.0	40.5 (5a)	>99	47.0 (3a)	>99	50	>200
2	2	Me	(\pm)- <i>trans</i> - 3b	12.5	40.0 (5b)	>99	45.0 (3b)	99	50	>200
3	1	allyl	(\pm)- <i>trans</i> - 4a	34.0	42.5 (6a)	>99	48.5 (4a)	99	50	>200
4	2	allyl	(\pm)- <i>trans</i> - 4b	30.0	42 (6b)	>99	47.5 (4b)	>99	50	>200

^a After purification by flash column chromatography on SiO₂. ^b Determined by chiral GC. ^c Calculated from the e.e. of the substrate and the product.⁸

TABLE 2. CAL-B-Catalyzed Enantioselective Acylation of (\pm)-*cis*- β -Hydroxy Nitriles

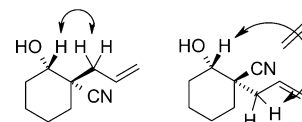
entry	<i>n</i>	R	substrate	<i>t</i> (h)	product		substrate		<i>c</i> ^c (%)	<i>E</i> ^c
					yield ^a (%)	ee (%) ^b	yield ^a (%)	ee ^b (%)		
1	1	Me	(\pm)- <i>cis</i> - 3a	4	44.5 (7a)	>99	44.0 (3a)	>99	50	>200
2	2	Me	(\pm)- <i>cis</i> - 3b	13	43.0 (7b)	>99	44.0 (3b)	99	50	>200
3	1	allyl	(\pm)- <i>cis</i> - 4a	8	35.0 ^d (8a)	>99	43.5 (4a)	>99	50	>200
4	2	allyl	(\pm)- <i>cis</i> - 4b	8	49.0 (8b)	>99	40.5 (4b)	>99	50	>200

^a After purification by flash column chromatography on SiO₂. ^b Determined by chiral GC. ^c Calculated from the ee of the substrate and the product.⁸ ^d Some of the product was lost during purification.

SCHEME 3

cis isomers (**4a** and **4b**), the CN signal is considerably deshielded in comparison to that of the corresponding *trans* isomer because of the γ steric-compression effect of the vicinal hydroxyl group.^{9b} As expected, the opposite effect is observed with the CH₂ group (see the Supporting Information). Also in the NOESY spectrum of compound *cis*-**4b** a correlation between the CH₂ group of the allyl substituent and the CH group of the cycle could be observed confirming the *cis* disposition of these two groups, whereas no correlation was observed between the same protons of the *trans*-**4b** diastereomer.

Regardless of the α -substituent and the ring size, CAL-B only reacted with substrates containing the alcohol with the absolute configuration *R*. In the case of hydroxy nitriles (1*S*,2*S*)-**3a, b** and (1*R*,2*S*)-**3a, b**, the absolute stereochemistry was assigned by comparison of the sign of their specific rotations with the published data.^{9a} The absolute configuration of the enzymatically prepared (1*R*,2*R*)-**6a,b** and (1*S*,2*R*)-**8a,b** has been tentatively assigned on the basis of two convergent criteria or by a double-confirmation method:¹⁰ (a) the enantiotopic preference displayed by CAL-B in the transesterification of secondary alcohols following in all cases the Kazlauskas's rule;¹¹ (b) using Mosher's method for secondary alcohols¹² to determine the *S*-configuration of the re-

**FIGURE 1.** Relevant NOESY correlations observed for (\pm)-*cis*-**4b** and (\pm)-*trans*-**4b**.

maining substrate (1*S*,2*S*)-**4b**. The differential shielding obtained for the α -methylene group of enantiopure Mosher derivative of (1*S*,2*S*)-**4b** and its enantiomer, ($\Delta\delta = +0.05$ ppm, signal at 1.98 ppm for the Mosher derivative of (1*S*,2*S*)-**4b** minus the signal at 1.93 ppm for its enantiomer in Mosher derivative of racemic (\pm)-*trans*-**4b**) obtained from the Mosher derivative of racemic (\pm)-*trans*-**4b** is in good agreement with previous findings. Where the positive sign of the $\Delta\delta$ values indicates that the product contained the alcohol (*S*)-configured¹⁰ (see the Supporting Information).

In summary, we have reported a novel resolution of five- and six-membered cyclic β -hydroxy nitriles bearing a quaternary center α to the nitrile utilizing lipase from *Candida antarctica* B as the biocatalyst for the enantioselective transesterification of these substrates. This procedure allows, for the first time, a direct access to all 16 stereoisomers of these interesting building blocks in enantiopure form and high yield.

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Supporting Information Available: ¹H and ¹³C NMR spectra of new compounds, experimental procedures, and experimental data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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